

## ANALYSIS ON NUMERICAL SIMULATIONS OF CO<sub>2</sub> ABSORPTION PROCESS OF CARBON SOLIDIFICATION SYSTEM

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**Abstract.** Carbon dioxide (CO<sub>2</sub>) emission control has been a popular topic since global warming affects our living conditions on the planet. Carbon Capture and Storage (CCS) is a feasible solution for mitigating the global warming effect by capturing the CO<sub>2</sub> from power stations and industrial processes and storing them underground. There are great many of active CCS projects onshore. International Maritime Organization (IMO) has adopted guidelines for CO<sub>2</sub> emission control by improving ship energy efficiency. The project is aiming to apply CCS on ships by capturing the CO<sub>2</sub> emission from the engine exhaust gases and solidifying them for easy storage. It will enable ships to comply with various regional and international CO<sub>2</sub> emission regulations and also maintain the efficiency of waterborne transportation. The simulation of carbon absorption develops and investigates the multiphase reaction model in CFD software, focusing on bubble column effect with chemical reaction. This paper presents the analysis of the numerical simulation of the CO<sub>2</sub> absorption process. Simulation results illustrate the pressure distribution of solution and gas path and velocity in solution. This numerical simulation results also indicates the impact of environment temperature on chemical reaction rate. A comparison between experiment and simulation results is presented to figure out the impact of initial NaOH solution concentration on gas absorption process. An optimized NaOH solution concentration is figured out and will be used for further practical design of absorption system for ships.

### REFERENCES

- [1] Metz, B., O. Davidson, H. C. de Coninck, M. Loos, and L. A. Meyer (eds.) (Working Group III of the Intergovernmental Panel on Climate Change), 2007, *IPCC Special Report on Carbon Dioxide Capture and Storage*, Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 442 pp.
- [2] World Resources Institute (WRI), 2008, *CCS Guidelines: Guidelines for Carbon Dioxide Capture, Transport, and Storage*, Washington, DC: WRI. ISBN 978-1-56973-701-9. [http://pdf.wri.org/ccs\\_guidelines.pdf](http://pdf.wri.org/ccs_guidelines.pdf)
- [3] Buhaug, Ø., Corbett, J.J., Endresen, Ø., Eyring, V., Faber, J., Hanayama, S., Lee, D.S., Lee, D., Lindstad, H., Markowska, A.Z., Mjelde, A., Nelissen, D., Nilsen, J., Pålsson, C., Winebrake, J.J., Wu, W., Yoshida, K., International Maritime Organization (IMO), 2009, *Second IMO GHG Study 2009*, London, UK, April 2009;.
- [4] Marine Environment Protection Committee (MEPC), 2011, *MARPOL Annex VI, Chapter IV*.
- [5] Global CCS Institute 2012, *the Global Status of CCS: 2012*, Canberra, Australia. ISBN 978-0-9871863-1-7. <http://decarboni.se/sites/default/files/publications/47936/global-status-ccs-2012.pdf>
- [6] DNV, 2013, *DNV and PSE report on ship carbon capture & storage*: [http://www.dnv.com/press\\_area/press\\_releases/2013/dnv\\_and\\_pse\\_report\\_on\\_ship\\_carbon\\_capture\\_storage.asp](http://www.dnv.com/press_area/press_releases/2013/dnv_and_pse_report_on_ship_carbon_capture_storage.asp)
- [7] Pflug, I.J., Angelini, P., Dewey, D.H., 1957, *Fundamentals of Carbon Dioxide Absorption as They Apply to Controlled-atmosphere Storage*, Department of Agricultural Engineering and Horticulture.
- [8] M., Mahmoudkhani, D.W., Keith, 2009. Low-energy sodium hydroxide recovery for CO<sub>2</sub> capture from atmospheric air—Thermodynamic analysis, Energy and Environmental System Group, Institute for Sustainable Energy, Environment, Economy, University of Calgary, Canada.
- [9] ANSYS Fluent 14.5, 2012, *ANSYS Fluent 14.5 Theory Guide*, USA. <http://www.ansys.com>